# Conjugation of Iron Oxide Nanoparticles with RGD-Modified Dendrimers for Targeted Tumor MR Imaging

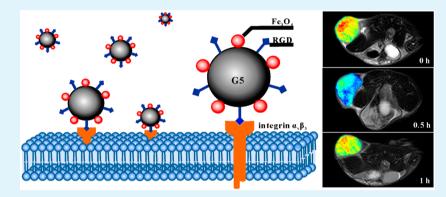
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# Supporting Information



**ABSTRACT:** This article reports a new approach for the synthesis of ultrasmall iron oxide nanoparticles (NPs) conjugated with Arg-Gly-Asp (RGD)-modified dendrimers (G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs) as a platform for targeted magnetic resonance (MR) imaging of C6 glioma cells. Ultrasmall Fe<sub>3</sub>O<sub>4</sub> NPs synthesized via a solvothermal route were conjugated with RGD peptide-modified generation-5 poly(amidoamine) dendrimers (G5.NH<sub>2</sub>-RGD). The final G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs were formed following the acetylation of the remaining dendrimer terminal amines. The as-prepared multifunctional Fe<sub>3</sub>O<sub>4</sub> NPs were characterized using various techniques. The results of a cell viability assay, cell morphological observation, and hemolysis assay indicated that the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs exhibit excellent cytocompatibility and hemocompatibility over the studied concentration range. In addition, RGD conjugated onto the Fe<sub>3</sub>O<sub>4</sub> NPs allows for the efficient targeting of the particles to C6 cells that overexpress  $\alpha_v\beta_3$  receptors, which was confirmed via *in vitro* cell MR imaging and cellular uptake. Finally, the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs offer significant potential for use as contrast agents in the targeted MR imaging of C6 glioma cells in mice. The results obtained from the current study indicate that the developed G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs offer significant potential for use as contrast agents in the targeted MR imaging of different types of tumors.

KEYWORDS: ultrasmall iron oxide nanoparticles, dendrimer, RGD, MR imaging, cancer cells

# INTRODUCTION

Recently, molecular imaging techniques have been widely developed for the diagnosis and detection of early stage tumors.<sup>1-4</sup> Among the current imaging modes, magnetic resonance (MR) imaging is a useful and noninvasive modality because of its high tomographic capabilities,<sup>5,6</sup> long effective imaging windows,<sup>7–9</sup> and high resolution and accuracy.<sup>10–12</sup> To improve the imaging sensitivity, selected contrast agents are important in clinical trials. Recent advances in nanotechnology have made it possible to use nanoparticles (NPs) as contrast agents for MR imaging. For example, Gd(III)- or Mn(II)-based NPs can be used for T<sub>1</sub>-weighted MR imaging,<sup>13–15</sup> and superparamagnetic iron oxide (Fe<sub>3</sub>O<sub>4</sub>) NPs can be employed as negative contrast agents for T<sub>2</sub>-weighted MR imaging

applications.<sup>16–19</sup> Because of their unique magnetic properties, high sensitivity, and low toxicity, superparamagnetic  $Fe_3O_4$  NPs have advantages over Gd(III)- or Mn(II)-based NPs as contrast agents for MR imaging.<sup>20,21</sup>

However, superparamagnetic  $Fe_3O_4$  NPs tend to aggregate and show a decrease in dissolution rates, especially if they are exposed to biological fluids or magnetic fields.<sup>22,23</sup> Therefore, appropriate surface modifications or coatings are necessary to improve the water dispersity, colloidal stability, and biocompatibility of  $Fe_3O_4$  NPs. Various polymers or small-molecule-

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coated  $\text{Fe}_3\text{O}_4$  NPs have been reported for different biomedical applications.<sup>24</sup>

Another issue associated with the use of  $Fe_3O_4$  NPs in tumor diagnosis is the lack of ability to target tumor sites. A targeting strategy could reduce the side effects of the particles on normal cells but increase their uptake by cancer cells. Therefore, it would be advantageous to construct  $Fe_3O_4$  NPs with specific targeting abilities after modification with certain ligands that can recognize molecular signatures on the surface of cancer cells, e.g., peptides,<sup>25–28</sup> folic acid (FA),<sup>2,29,30</sup> antibodies,<sup>31,32</sup> galactose,<sup>33,34</sup> chlorotoxin,<sup>35,36</sup> and hyaluronic acid (HA).<sup>37–39</sup>

In a previous study, ultrasmall  $Fe_3O_4$  NPs were successfully synthesized via a simple one-pot solvothermal method using trisodium citrate as a crystal grain growth inhibitor and stabilizer.<sup>40</sup> The as-synthesized  $Fe_3O_4$  NPs, which exhibited a range of sizes, not only showed excellent colloidal stability and good biocompatibility but also displayed significant potential for use as  $T_1$  contrast agents.

In this study, ultrasmall Fe<sub>3</sub>O<sub>4</sub> NPs synthesized via the above-described method (Fe<sub>3</sub>O<sub>4</sub>-COOH NPs) were further conjugated with Arg-Gly-Asp (RGD)-modified generation-5 poly(amidoamine) dendrimers (G5.NH<sub>2</sub>-RGD). Following acetylation, ultrasmall Fe<sub>3</sub>O<sub>4</sub> NPs conjugated with RGDmodified dendrimers (G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs) were formed. The products were characterized using various techniques, and the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs showed good water dispersibility, colloidal stability, and excellent hemocompatibility and biocompatibility over the studied concentration range. In vitro cellular uptake analysis suggested that G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs have a higher binding affinity toward C6 cells that overexpress  $\alpha_v \beta_3$  receptors than G5.NH.Ac-Fe<sub>3</sub>O<sub>4</sub> NPs. In vitro cell and in vivo tumor MR imaging experiments not only indicated good T2-weighted MR imaging performance but also demonstrated the specific targeting ability of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs to C6 cells; therefore, these nanoparticles can be used for targeted MR imaging of tumors via the RGD-mediated pathway.

### EXPERIMENTAL SECTION

Materials. Ethylenediamine core G5.NH2 PAMAM dendrimers with a polydispersity index less than 1.08 were purchased from Dendritech (Midland, MI, USA). The cyclic RGD peptide ( $M_w$  = 706.8) was purchased from GenicBio BioTech Co. Ltd. (Shanghai, China). Oleylamine, 1-ethyl-3-[3-(dimethylamino)propyl] carbodiimide hydrochloride (EDC), and N-hydroxysuccinimide (NHS) were supplied by J&K Chemical Ltd. (Shanghai, China). Ferric chloride hexahydrate (FeCl<sub>3</sub>·6H<sub>2</sub>O > 99%), diethylene glycol (DEG), trisodium citrate, triethylamine, acetic anhydride, and all other chemicals and solvents were purchased from Sigma-Aldrich (St. Louis, MO, USA) and used as received. C6 cells (a rat C6 glioma cell line) were obtained from the Institute of Biochemistry and Cell Biology, Chinese Academy of Science (Shanghai, China). RPMI 1640 medium, fetal bovine serum (FBS), penicillin, and streptomycin were obtained from Hangzhou Jinuo Biomedical Technology Co. Ltd., China, and 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) was purchased from Sigma-Aldrich. The water used in all experiments was purified using a Milli-Q Plus 185 water purification system (Millipore, Bedford, MA, USA) and exhibited a resistivity greater than 18.2 MQ·cm. Regenerated cellulose dialysis membranes (MWCO = 1,000) were acquired from Fisher.

Synthesis of  $Fe_3O_4$ -COOH NPs. Anhydrous  $FeCl_3$  (2 mmol) was first vigorously mixed with 20 mL of DEG to form a clear solution. Then, Na<sub>3</sub>Cit (0.8 mmol) was added, and the mixture was heated to 80 °C with stirring to form a clear solution. After dissolving anhydrous NaOAc (6 mmol) in the previously mentioned solution, the mixture was transferred to a Teflon-lined stainless-steel autoclave (50 mL) and sealed in air. Following this step, the autoclave was held at 200  $^{\circ}$ C for 4 h. After being naturally cooled to room temperature, the black product was collected by sedimentation with the aid of an external magnetic field and purified with ethanol three times to remove excess reactants and byproducts. The resulting precipitate was dried at 60  $^{\circ}$ C for further use.

**Fabrication of G5.NH<sub>2</sub>-RGD.** The G5.NH<sub>2</sub>-RGD conjugate was synthesized according to our previous report.<sup>15</sup> Briefly, G5.NH<sub>2</sub> (26.00 mg, 10 mL in DMSO) was first activated with 1 mol equiv of *N*-succinimidyl 6-maleimidohexanoate (6 M, 3.08 mg, 2 mL in DMSO) under vigorous magnetic stirring at room temperature for 8 h. Then, RGD (6.92 mg, 2 mL in DMSO) was added, and the ensuing reaction was allowed to proceed for 12 h, yielding a raw product of G5.NH<sub>2</sub>-RGD. The DMSO and excess reactants and byproducts were removed from the mixture via extensive dialysis against PBS (three times, 2 L) and water (six times, 2 L) for 3 d, followed by lyophilization to obtain G5.NH<sub>2</sub>-RGD.

Synthesis of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs. The carboxyl groups of Fe<sub>3</sub>O<sub>4</sub>-COOH NPs (56 mg, 15 mL HEPES buffer solution) were activated with EDC (144 mg) and NHS (70 mg) in MES buffer solution. Then, the activated Fe<sub>3</sub>O<sub>4</sub>-COOH NPs were added dropwise into the solution of G5.NH<sub>2</sub>-RGD (1  $\mu$ mol, 15 mL of MES buffer solution), and the mixture was stirred for 3 d. The remaining amino groups on the G5.NH<sub>2</sub>-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs were further acetylated to reduce their surface potentials. Briefly, 287  $\mu$ L of triethylamine was added to the G5.NH2-RGD-Fe3O4 NPs aqueous solution, and the solution was thoroughly mixed for 30 min. Then, 146  $\mu$ L of acetic anhydride was added dropwise into the above-mentioned solution under vigorous magnetic stirring at room temperature, and the reaction was allowed to proceed for 24 h. After being centrifuged and washed with water (at least three times), the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs were finally obtained and dispersed in 10 mL of water. As a control, G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs were also prepared using a similar route.

Characterization Techniques. <sup>1</sup>H NMR spectra were collected on a Bruker AV400 nuclear magnetic resonance spectrometer in DMSO- $d_6$ . MALDI-TOF mass spectra were acquired using a 4800 Plus MALDI TOF/TOFTM Analyzer (AB SCIEX, Framingham, MA) in linear mode. Thermogravimetric analysis (TGA) was performed using a TG 209 F1 (NETZSCH Instruments Co., Ltd., Germany) thermogravimetric analyzer, and the samples were heated from room temperature to 700 °C at a rate of 20 °C/min under nitrogen gas. The zeta potentials and hydrodynamic sizes of the particles were measured with a Malvern Zetasizer Nano ZS model ZEN3600 (Worcestershire, U.K.) equipped with a standard 633 nm laser. The sizes and morphologies of the particles were characterized using transmission electron microscopy (TEM, 200 kV, JEOL 2010F, Japan); the TEM sample was prepared by transferring one drop of solution onto a carbon-coated copper grid, which was left undisturbed until it had dried. The Fe concentration in all experiments was analyzed using a Leeman Prodigy Inductively Coupled Plasma-Atomic Emission Spectroscopy (ICP-AES) system (Hudson, NH03051, USA). T<sub>2</sub> relaxometry measurements and T2-weighted MR imaging were performed using an NMI20-Analyst NMR Analyzing and Imaging system (Shanghai Niumag Corporation). The samples were diluted with distilled water to different Fe concentrations, and the instrumental parameters were a magnetic field strength of 0.5 T, point resolution = 156 mm × 156 mm, section thickness = 0.6 mm, TR = 4000 ms, TE = 60 ms, and number of excitations = 1.

**Hemolysis Assay.** Fresh human blood stabilized with EDTA was kindly provided by Shanghai First People's Hospital (Shanghai, China). The fresh human blood was centrifuged to remove the serum, and the resulting human red blood cells (HRBCs) were purified by five successive repetitions of rinsing with PBS. After 10-fold dilution of the HRBC in PBS, aliquots of the HRBC suspension (0.1 mL) were added into Eppendorf tubes containing 0.9 mL of water (positive control), 0.9 mL of PBS (negative control), or 0.9 mL of PBS containing G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs at different Fe concentrations (5–100  $\mu$ g/mL). The mixtures were vortexed and held still for 2 h at

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room temperature. The samples were then centrifuged (10000 rpm, 1 min), photographs of the samples were taken, and the absorbance of the supernatants (hemoglobin) were collected using a PerkinElmer Lambda 25 UV–vis spectrometer. The hemolysis percentages of the different samples were calculated by dividing the difference in the absorbances between the samples and the negative control by the difference in absorbance between the positive and negative controls.

Cytotoxicity Assay and Cell Morphology Observation. C6 cells were continuously cultured in 25 cm<sup>2</sup> plates in regular RPMI 1640 medium supplemented with 10% heat-inactivated fetal bovine serum (FBS), penicillin (100 U/mL), and streptomycin (100  $\mu$ g/mL) at 37 °C and 5% CO2. The cytotoxicity evaluations were performed using a standard MTT assay. Briefly, C6 cells at an initial density of 1  $\times$  10<sup>4</sup> cells per well were seeded into a 96-well plate in 200  $\mu$ L of RPMI 1640 medium. After incubation for 12 h, the medium was replaced with 200  $\mu$ L of fresh medium containing pure PBS (control), G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs, or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs at different Fe concentrations (5, 10, 25, 50, and 100  $\mu$ g/mL). The MTT solution (20  $\mu$ L, 5 mg/mL in PBS) was added into each well to detect metabolically active cells after incubation at 37 °C and 5% CO2 for 24 h. The cells were further incubated at 37 °C for another 4 h, and 200  $\mu$ L of DMSO was added to each well to replace the culture medium. After dissolving the insoluble formazan crystals by mildly shaking the plate for 15 min, the absorbance of each well was measured using a Thermo Scientific Multiskan MK3 ELISA reader (Thermo Scientific, USA) at 570 nm. Each absorbance measurement was performed five times to calculate the mean and standard deviation.

The cell morphologies were also observed via phase contrast microscopy (Leica DM IL LED inverted phase contrast microscope) after the C6 cells were treated with PBS, G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs, or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs at different Fe concentrations (5, 10, 25, 50, and 100  $\mu$ g/mL) for 24 h. The magnification was set at 200× for each sample.

Cellular Uptake Assay. The cellular uptake of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs by C6 cells was quantitatively studied using ICP-AES. Briefly, C6 cells were seeded in 12-well plates at an initial density of  $5 \times 10^5$  cells/well with 1 mL of RPMI 1640 medium. After incubation at 37 °C and 5% CO<sub>2</sub> for 12 h, the medium was carefully discarded, and 1 mL of fresh medium containing G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs at different Fe concentrations (5, 10, 25, 50, and 100  $\mu$ g/mL) was added into each well. After incubation for 4 h, the cells were washed with PBS, harvested, and suspended in 1 mL of PBS. The cells were counted using a hemocytometer. Before the ICP-AES experiments, the cells were dissolved in an aqua regia solution and diluted twice with PBS. Three samples were collected under each condition to obtain the mean and standard deviation. In addition, the cellular uptake of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs by normal cells (L929 cells) was confirmed using ICP-AES and a method similar to the one described above.

*In Vitro* **Targeted Cell MR Imaging.** C6 cells were seeded into 6well plates at a density of  $2 \times 10^6$  cells per well with 2 mL of RPMI 1640 medium for 12 h. Then, the medium was replaced with 2 mL of fresh medium containing G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs at different Fe concentrations (5, 10, 25, 50, and 100  $\mu$ g/ mL), and the cells were sequentially incubated at 37 °C and 5% CO<sub>2</sub> for 4 h. The cells were then washed with PBS, harvested, and suspended in 1 mL of PBS (containing 0.5% agarose) in 2 mL Eppendorf tubes. T<sub>2</sub> MR imaging of the cell suspensions in each tube was performed using a GE Discovery MR750 3T scanner and an animal coil suitable for mouse imaging (Magtron Inc., Shanghai, China) with a T<sub>2</sub>-weighted FSE sequence (slice thickness of 3.4 mm, TR/TE 3000/87.4 ms, 8 × 8 cm FOV, and 320 × 256 matrix).

*In Vivo* Tumor Targeted MR Imaging. All animal experiments were performed according to protocols approved by the institutional committee for animal care and in accordance with the policy of the National Ministry of Health. Male BALB/c nude mice aged 4 to 6 weeks (15–20 g) were purchased from Shanghai SLAC Laboratory Animal Center (China). To establish the tumor model,  $2 \times 10^6$  C6 cells/mouse were subcutaneously implanted into the left back of the

nude mice. The experiments were performed when the xenografted tumor volume reached approximately 0.6 cm<sup>3</sup>. The C6 tumor-bearing nude mice were anesthetized by intraperitoneal injection of pentobarbital sodium (40 mg/kg). Subsequently, 0.1 mL of a PBS solution of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs (600  $\mu$ g Fe) was delivered into the C6 tumor-bearing nude mice via the tail vein. MR imaging was performed using a GE Discovery MR750 3T scanner with an animal coil suitable for mouse imaging (Magtron Inc., Shanghai, China). The mice were imaged before and after administration of the particles at 0.5, 1, 2, and 4 h. The parameters were 1.8 mm slice thickness, TR/TE 4000/96.6 ms, 3 × 3 cm FOV, and 320 × 256 matrix.

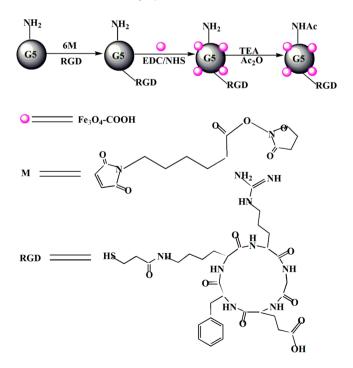
*In Vivo* Biodistribution. The C6 tumor-bearing nude mice were submitted to *in vivo* biodistribution analysis to understand the metabolic behavior of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs. The mice were first anesthetized via intraperitoneal injection of pentobarbital sodium (40 mg/kg). Then, 0.1 mL of a PBS solution of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs (600  $\mu$ g Fe) was intravenously injected into the C6 tumor-bearing nude mice. At different times, the mice were euthanized, and the main organs, i.e., the heart, liver, spleen, lungs, and kidneys, and the tumor tissue were extracted and weighed. For the ICP-AES measurements, these tissues were subsequently cut into 1–2 mm<sup>2</sup> pieces and digested in aqua regia solutions for 1 d.

## RESULTS AND DISCUSSION

Synthesis and Characterization of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs. In this study, citric acid-stabilized ultrasmall Fe<sub>3</sub>O<sub>4</sub>-COOH NPs were successfully synthesized according to previous work. RGD-conjugated generation-5 poly-(amidoamine) dendrimers (G5.NH<sub>2</sub>-RGD) were then used as templates for synthesizing the targeting particles, followed by acetylation of the remaining dendrimer terminal amines (Scheme 1). The resulting G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs were characterized using various techniques.

RGD-conjugated G5.NH<sub>2</sub> was synthesized using 6-maleimidohexanoic acid *N*-hydroxysuccinimide ester as a heterobifunctional cross-linker according to previous work, and the yield of G5.NH<sub>2</sub>-RGD was 86.3%. After purification via dialysis, the

Scheme 1. Schematic Representation of the Preparation of the G5.NHAc-RGD-Fe $_3O_4$  NPs



resulting G5.NH<sub>2</sub>-RGD was characterized using MALDI-TOF mass spectroscopy (Figure S1, Supporting Information). In the MALDI-TOF mass spectra, the  $M_w$  of G5.NH<sub>2</sub> (26095) increased to 27425 after RGD conjugation, suggesting that 2 RGD molecules were coupled to each G5.NH<sub>2</sub>.

TGA was performed to analyze the inorganic content of the different particles (Figure 1). The Fe<sub>3</sub>O<sub>4</sub>-COOH NPs showed a

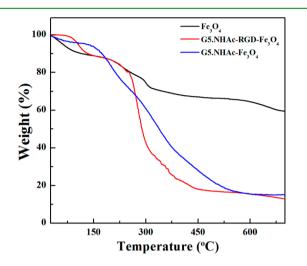


Figure 1. TGA curves of  $Fe_3O_4$ -COOH NPs, G5.NHAc-Fe $_3O_4$  NPs, and G5.NHAc-RGD-Fe $_3O_4$  NPs.

weight loss of 68.6% due to the citric acid surface coating of the particles. After modification with G5.NH<sub>2</sub> or G5.NH<sub>2</sub>-RGD and acetylation, the weight loss was 16.8% for G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and 16.9% for G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs. The reduction in the inorganic content of the particles suggested that the chemical reaction and acetylation were successfully achieved.

Zeta potential measurements and dynamic light scattering (DLS) were employed to confirm the surface charge and hydrodynamic sizes of the different products. As shown in Table 1,  $Fe_3O_4$ -COOH NPs, G5.NHAc-Fe\_3O<sub>4</sub> NPs, and

Table 1. Zeta Potentials and Hydrodynamic Sizes of  $Fe_3O_4$ -COOH, G5.NHAc-Fe\_3O<sub>4</sub>, and G5.NHAc-RGD-Fe\_3O<sub>4</sub> NPs

sample	zeta potential (mV)	hydrodynamic size (nm)
Fe <sub>3</sub> O <sub>4</sub> -COOH	-43.4	18.6
G5.NHAc-Fe <sub>3</sub> O <sub>4</sub>	-4.8	453.9
G5.NHAc-RGD-Fe <sub>3</sub> O <sub>4</sub>	-5.5	527

G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs have surface potentials of -43.4 mV, -4.8 mV, and -5.5 mV, respectively, due to the different functional groups on the surface of the respective particles. The hydrodynamic size results are also shown in Table 1; the hydrodynamic size of the G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs (453.9 nm) and G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs (527.0 nm) was much larger than that of the Fe<sub>3</sub>O<sub>4</sub>-COOH NPs (18.6 nm). This result suggests that clusters might have formed after the Fe<sub>3</sub>O<sub>4</sub>-COOH NPs were modified with G5.NHAc-RGD.

The morphology and size of the G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs were characterized using TEM. It can be clearly observed that the Fe<sub>3</sub>O<sub>4</sub>-COOH NPs have a very small spherical shape and a quite uniform size distribution (Figure S2, Supporting Information). After modification, the G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs showed

cluster structures, i.e., some interconnected particles appeared (Figure 2).

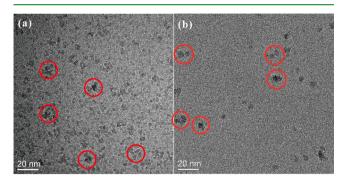
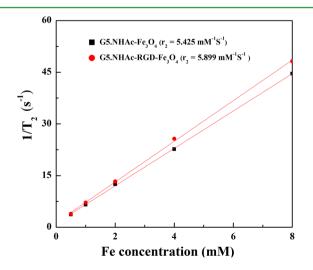


Figure 2. TEM images of G5.NHAc-Fe $_3O_4$  NPs (a) and G5.NHAc-RGD-Fe $_3O_4$  NPs (b).

The G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs were dispersed in different media, i.e., water, PBS, and cell culture medium, to assess their colloidal stability (Figure S3, Supporting Information). Photographs showed that the particles were stable and that no precipitates occurred in these media after 1 d.

Transverse Relaxivity  $(r_2)$  Measurement. Ultrasmall Fe<sub>3</sub>O<sub>4</sub> NPs with diameters less than 5 nm have been widely used as contrast agents for T<sub>1</sub> MR imaging. Additionally, these particles exhibit a T2-weighted MR imaging effect if certain clusters of particles are formed, as demonstrated in this study. The T<sub>2</sub> values of the water protons in an aqueous solution of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs at different Fe concentrations were measured at 0.5 T, and the transverse relaxivities  $(r_2)$  of the particles were calculated based on the measured T<sub>2</sub> values. The results showed that the relaxation rate increased linearly with the Fe concentration for both types of particles and that the  $r_2$  values of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs were calculated to be 5.899 mM<sup>-1</sup> s<sup>-1</sup> and 5.425 mM<sup>-1</sup> s<sup>-1</sup>, respectively (Figure 3). In addition, we measured the longitudinal relaxivity  $(r_1)$  of the particles (Figure S4, Supporting Information). It can be seen that G5.NHAc-RGD-Fe3O4 and G5.NHAc-Fe3O4 NPs have  $r_1$  values of 0.62 m $M^{-1}$  s<sup>-1</sup> and 0.66 m $M^{-1}$ 



**Figure 3.** Linear fitting of  $1/T_2$  for G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs with Fe concentrations of 0.5,1, 2, 4, and 8 mM.

respectively. The large  $r_2/r_1$  ratios (9.51 for the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and 8.22 for the G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs) suggest that these particles have excellent potential to be used as contrast agents for T<sub>2</sub>-weighted MR imaging. The T<sub>2</sub>-weighted images of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs are shown in Figure 4. The particles induced a

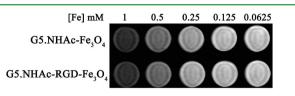
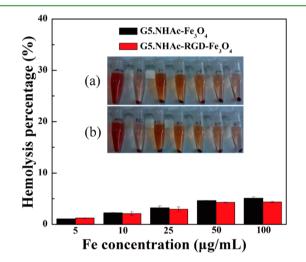


Figure 4.  $T_2$ -weighted images of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs with Fe concentrations of 0.0625, 0.125, 0.25, 0.5, and 1 mM.

decrease in signal over the studied concentration range (0.0625–1 mM), indicating that the NPs synthesized in this study could be used as contrast agents for  $T_2$  MR imaging.

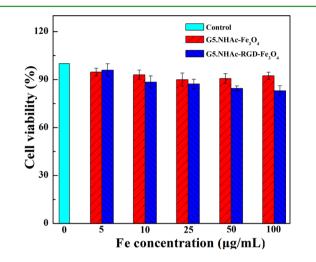
**Hemolysis Assay.** It is very important for NPs to exhibit good hemocompatibility for use in various *in vivo* biomedical applications. The hemocompatibility of the developed G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs was evaluated using a hemolytic assay (Figure 5). On the basis of



**Figure 5.** Hemolytic activity of G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs (a) and G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs (b) at different Fe concentrations (5, 10, 25, 50, and 100  $\mu$ g/mL). PBS and water were used as the negative and positive controls, respectively. The middle insets show photographs of HRBCs exposed to water, PBS, and PBS containing different concentrations of NPs for 2 h followed by centrifugation.

the photographs captured, no distinct hemolysis could be observed for either particle type over the studied concentration range compared with the effects of water (positive control) or PBS (negative control). Quantitative analyses show that the hemolysis percentages of these particles were all less than 5%, even when the Fe concentration reached 100  $\mu$ g/mL. Our results suggest that the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs exhibit excellent hemocompatibility over the studied concentration range.

**Cytotoxicity Assay and Cell Morphology Observation.** Prior to implementing the nanoparticles in *in vivo* applications, it is necessary to explore the cytotoxicity of the developed G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs *in vitro*. After incubating C6 cells with these particles at Fe concentrations of 5, 10, 25, 50, and 100  $\mu$ g/mL for 24 h, an MTT assay was performed using the standard protocol (Figure 6). No significant differences in the viabilities of the cells



**Figure 6.** MTT assay of C6 cell viability after treatment with G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs at Fe concentrations of  $0-100 \ \mu g/mL$  for 24 h. C6 cells treated with PBS were used as the control.

treated with the particles were observed over the studied Fe concentration range of 5–100  $\mu$ g/mL compared with the viability of the PBS control. This MTT result suggests that the developed G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs exhibit remarkably low cytotoxicities over the given concentration range.

The morphologies of the C6 cells treated with different concentrations of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs over 24 h were observed using phase contrast microscopy to further confirm the particles' lack of cytotoxicity (Figure S5, Supporting Information). All of the treated cells showed morphologies similar to those of the control cells treated with PBS, further suggesting that the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs have good cytocompatibility.

Cellular Uptake Assay. ICP-AES was used to study the cellular uptake of the different particles by C6 cells (Figure 7). Concentration-dependent uptake was observed for both the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and the G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs. However, at the same Fe concentration, the uptake of the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs by the C6 cells was higher than that of the G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs, particularly when the Fe concentration reached 100  $\mu$ g/mL. The enhanced uptake could be due to the high affinity of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs for  $\alpha_{s}\beta_{3}$  receptors, which are overexpressed on the surfaces of C6 cells. In sharp contrast, at the same concentration, we can clearly see that G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs have increased uptake by L929 cells; however, there is no obvious difference between the two different particles at similar concentrations (Figure S6, Supporting Information). It should be noted that the cellular uptake of the G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs by C6 cells and G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs by L929 cells may be attributed to two distinct mechanisms (phagocytosis and diffusion through cell walls) and that the enhanced cellular uptake of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs by C6 cells is due to the receptor-mediated pathway.<sup>2</sup>

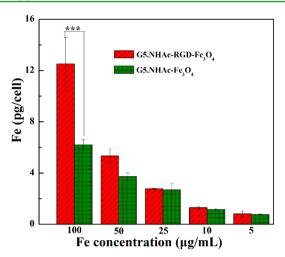


Figure 7. In vitro cellular uptake of G5.NHAc-RGD-Fe $_3O_4$  NPs and G5.NHAc-Fe $_3O_4$  NPs by C6 cells after treatment with different Fe concentrations for 4 h.

In Vitro Targeted Cell MR Imaging. To further confirm the special ability of the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs to target C6 cells and to validate the particles' cell MR imaging performance, *in vitro* targeted cell MR imaging was performed. The T<sub>2</sub>-weighted MR images of the treated cells showed a significant signal drop as a function of Fe concentration (Figure 8a). Additionally, at the same Fe concentration, the signal of

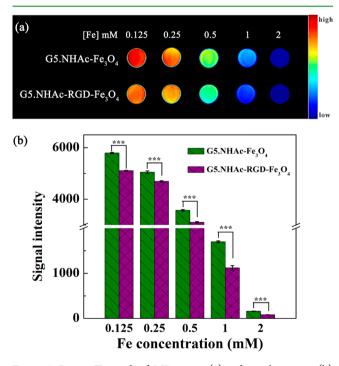


Figure 8. In vitro  $T_2$ -weighted MR images (a) and signal intensity (b) of C6 cells treated with G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs at different Fe concentrations for 4 h.

the C6 cells treated with G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs was much lower than that of the same cells treated with G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs. By plotting the T<sub>2</sub> MR signal intensity of the C6 cells treated with G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs as a function of Fe concentration (Figure 8b), it can be clearly observed that for both types of particles, the MR signal intensity decreases with the Fe concentration and that the signal intensity from the cells treated with G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs is clearly lower than that of similar cells treated with G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs at the same Fe concentration. This result further demonstrates that the modified RGD can direct the particles' targeting of C6 cells with high  $\alpha_{\rm v}\beta_3$  receptor expression.

In Vivo Targeting MR Imaging of Tumors. The promising in vitro MR imaging and targeting cellular uptake results prompted an investigation into the application of G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs for in vivo targeted MR imaging of C6 tumor-bearing mice. Figure 9a shows T<sub>2</sub>-weighted MR images of a tumor obtained before injection and 0.25, 0.5, 1, and 2 h postinjection. The tumor MR signal from the mice injected with G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs decreased from 0.25 to 0.5 h compared with the signal obtained for mice that were not injected. In contrast, no significant signal change was observed for the tumors in mice treated with G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs during this period. One hour after injection, the MR signals of the two tumor sites were recovered, indicating that the injected particles were ejected from the tumors via a further metabolic process. The MR signal intensities from the tumors of the mice were quantitatively analyzed as a function of the postinjection time and the treatment (Figure 9b). For the targeted group, the tumor signal intensity decreased significantly from 9020 (0 h) to 4793 at 0.5 h postinjection. In contrast, only a slight decrease in MR signal intensity could be observed in the control group (mice treated with G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs). This result suggests that in addition to the passive enhanced permeability and retention (EPR) effect, the RGDmediated targeting pathway plays an important role in the uptake of G5.NHAc-RGD-Fe3O4 NPs. One hour after injection, the tumor MR signal intensity of the two groups increased due to further metabolic processes. Our results suggest that the developed G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs have good potential for use as a tumor-targeted negative contrast agent for T<sub>2</sub> MR imaging.

*In Vivo* Biodistribution. To explore the biodistribution behavior of the injected G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs, ICP-AES was used to analyze the Fe contents in the tumor tissue and the major organs, i.e., the liver, spleen, lungs, and kidneys, at different post-treatment times (Figure 10). At 12 and 24 h, Fe was primarily taken up by the liver, spleen, and lung, and only a small amount of Fe remained in the tumor tissue and other organs, i.e., the heart and kidneys. At 48 h, the Fe content in the liver, spleen, and lung decreased due to further metabolic processes. Our results suggest that the injected G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs can be successfully cleared from the body and will not have negative effects on animals.

# CONCLUSIONS

In summary, RGD-targeted ultrasmall Fe<sub>3</sub>O<sub>4</sub> NPs were developed for targeted T<sub>2</sub>-weighted MR imaging of C6 cells *in vitro* and xenografted C6 tumors *in vivo* through a dendrimermediated approach. Our results indicate that the targeted NPs show good water and colloidal stability, a relaxivity ( $r_2$ ) of 5.899 mM<sup>-1</sup> s<sup>-1</sup>, and good hemocompatibility and biocompatibility over the studied concentration range. Cellular uptake studies, *in vitro* cell MR imaging, and *in vivo* tumor imaging confirm that the developed G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs have a high affinity for C6 cells that overexpress  $\alpha_v \beta_3$  receptors. The findings obtained in this study suggest that the developed G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs have as contrast agents for targeted T<sub>2</sub> MR imaging of specific tumors.

#### Research Article

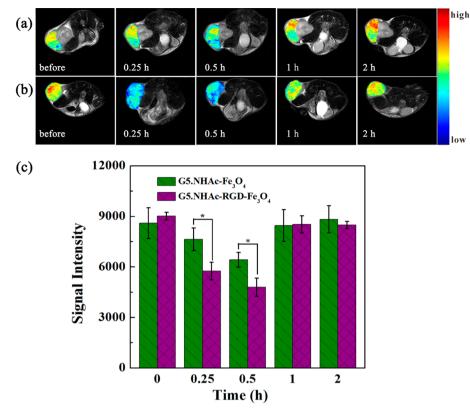
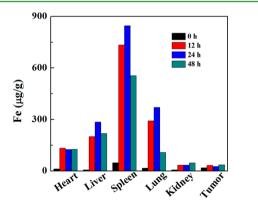


Figure 9. In vivo  $T_2$ -weighted MR images of tumors (a, b) and the signal intensity (c) after intravenous injections of 0.1 mL of PBS solutions containing G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs (a) or G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs (b) (600  $\mu$ g Fe) at different times.



**Figure 10.** Biodistribution of the major organs and tissue of the mice, i.e., heart, liver, spleen, lung, kidney, and tumor, at different times post-intravenous injection of 0.1 mL of a PBS solution containing G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs (600  $\mu$ g Fe).

# ASSOCIATED CONTENT

#### **S** Supporting Information

MALDI-TOF mass spectra of G5.NH<sub>2</sub> and G5.NH<sub>2</sub>-RGD dendrimers; TEM images and size distribution histogram of Fe<sub>3</sub>O<sub>4</sub>-COOH NPs; photographs of G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs dispersed in water, PBS, and culture medium; linear fitting of  $1/T_1$  for the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs; phase-contrast microscopy images of C6 cells treated with PBS or G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs; and *in vitro* cellular uptake of the G5.NHAc-RGD-Fe<sub>3</sub>O<sub>4</sub> NPs and G5.NHAc-Fe<sub>3</sub>O<sub>4</sub> NPs by L929 cells. This material is available free of charge via the Internet at http://pubs.acs.org.

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## Author Contributions

<sup>II</sup>J.Y., Y.L., and Y.X. contributed equally to this work.

#### Notes

The authors declare no competing financial interest.

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